

Automated Image Analysis of Mineral Matter  
in Raw and Supercleaned Coals

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The advent of highly sophisticated and automated microscopic techniques, together with powerful mini- and microcomputers, makes it now possible to characterize the mineral components of coal in-situ. Combined automated image analysis (AIA) and scanning electron microscopy (SEM) allows detailed characterization of minerals in coal for size, shape, composition, and relation to the coal matrix. For a statistically significant number of particles, both size distribution and volume fraction can be estimated and used to characterize independently the mineral matter content. The corresponding chemical analyses for the bulk coal samples, provided by more conventional techniques, may be related to the AIA-SEM data for comparison.

Conventional analytical techniques such as x-ray diffraction (XRD) and Fourier transform infrared spectroscopy (FTIR) can identify the mineral phases present in coal and can sometimes provide an estimation of the amount present. However, such techniques use bulk samples; they are limited to calculating only an "average" cleaning effectiveness and do not offer information on the size distribution of those mineral phases identified. On the other hand, AIA-SEM permits cleaning effectiveness to be evaluated with respect to both particle size and mineral phase. Thus, problems associated with removing a particle size or chemical class of particles can be detected and addressed. Such information is important for any coal preparation process, especially since it relates grinding and liberation of mineral particles to washability tests (1).

To evaluate coal particles produced by fine grinding for washability tests, the AIA-SEM technique provides information on the elemental distribution among the various mineral phases. The minerals are classified using a chemistry definition file based on the relative amounts of elements present as determined by energy-dispersive x-ray spectroscopy.

The AIA-SEM technique has been used in Ames Laboratory to study the effect of grinding on washability of fine coal. Several series of coals have been characterized by this technique in the past two years, thus demonstrating its usefulness. In this work, the AIA-SEM technique was applied to determine the coal mineral character before and after cleaning.

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## EXPERIMENTAL

### Sample Description and Preparation

The analyses were performed on two bituminous coals being tested for washability by conventional means. The coals were from the Illinois No. 6 seam, Randolph county, Illinois, and from the Pittsburgh No. 8 seam, Lewis county, West Virginia. The coals were ground to a typical power plant grind (i.e., 70-80% less than 200 mesh or 75  $\mu\text{m}$ ). The coals were then supercleaned by float-sink separation (using halogenated hydrocarbons) at 1.3 specific gravity to produce a very low-ash, clean coal fraction (ash content <3%). The raw and clean coal fractions were analyzed for moisture, ash, and sulfur forms by the usual ASTM procedures (see Table 1).

Table 1. ASTM analyses of raw and supercleaned coal samples for moisture, ash, forms of sulfur, and mineral matter<sup>a</sup>.

	Illinois No. 6		Pittsburgh No. 8	
	Raw	Clean	Raw	Clean
Moisture	16.90	1.95	1.97	1.12
Ash	16.11	2.61	6.75	2.97
Total S	5.10	2.54	3.17	1.82
Pyritic S	2.37	0.22	1.35	0.03
Sulfate S	0.36	0.04	0.41	0.12
Organic S	2.36	2.27	1.42	1.67
Mineral Matter <sup>b</sup>	19.32	3.05	8.26	3.37

<sup>a</sup> Values are expressed as wt. % on a dry basis, except for moisture.

<sup>b</sup> Mineral matter =  $1.13 \times \text{ash} + 0.47 \times (\text{pyritic sulfur})$ .

The raw and clean coal samples were prepared for AIA-SEM analyses by casting two grams of the sample with an epoxy resin into molds one inch in diameter. The hardened pellets were polished to reveal a cross section, with final polishing being done with 0.3  $\mu\text{m}$  alumina powder. The pellets were coated with approximately 500 angstroms of carbon to render sample surfaces electrically conductive for examination in the electron microscope.

### AIA-SEM Analysis

The AIA-SEM system consists of a JEOL (Japan Electron Optics Laboratory) model JSM-U3 scanning electron microscope, a LeMont Scientific B-10 image analyzer, and a Tracor Northern TN-2000 energy-dispersive x-ray analyzer. The image analyzer is a software-based system with associated electronics for SEM beam control, image amplification, and thresholding. The software base for image analysis allows the appropriate analysis algorithm to be selected for the particular sample and image conditions encountered. Particle extents are determined from the

points at which horizontal scans cross a feature, and particle outline is reconstructed from these adjacent chords of a particle. Once the outline has been determined, the x-ray data are collected from the center of the particles.

Samples were analyzed in the SEM using 25 kV beam voltage, 1-2 nA sample currents, 300x magnification, and backscattered electron imaging. A point density of 1024 pixels across the screen was used to provide  $\pm 10\%$  accuracy on measurements as small as 1% of the field of view. X-ray data were collected for four seconds per particle at a typical counting rate of 1000 counts per second. Regions of interest were set to monitor the intensities of 30 elements; however, only these 11 elements occurred with significant frequency: Na, Mg, Al, Si, P, S, Cl, K, Ca, Ti, and Fe. Approximately 4000 particles were analyzed per sample at a rate of 200 particles per hour.

#### Data Handling

Based on the relative amounts of the elements present, the particles were classified into one of nine mineral categories according to the definitions given in Table 2. The categories derived from previously established guidelines (2,3) and included the common coal minerals pyrite, quartz, calcite, siderite, kaolinite, and illite. Several other minerals were identified, but they occurred in such small amounts that they were classified together into one common category titled "MINORS". This category included the minerals gypsum, dolomite,

Table 2. Chemical limits for mineral phase definitions

Mineral Phase	Chemical Definition in % Range <sup>a</sup>	Specific Gravity
PYRITE	S 10-80; Fe 10-70	5.00
KAOLINITE	Al 15-80; Si 15-85; Al/Si 0.33-3.0	2.65
ILLITE	Al 10-50; Si 20-85; Mg 0-15; Ca 0-35; Fe 0-40; Ti 0-15	2.75
QUARTZ	Si 65-100	2.65
CALCITE	Ca 70-100	2.80
SIDERITE	Fe 70-100; Mn 0-30; Ni 0-30	5.00
MINORS (includes the following categories)		
GYPSUM	S 10-80; Ca 10-70	2.30
DOLOMITE	Mg 10-60; Ca 60-100	2.90
RUTILE	Ti 70-100	4.50
ALUMINA	Al 65-100	4.00
APATITE	P 15-40; Ca 30-100	3.20
SILICATES	Si 20-80	2.70
MISCELLANEOUS	(no restrictions, all particles accepted)	2.00

<sup>a</sup>Specifications may be given for the amount of other elements that are allowed to be present. Such specifications allow minor amounts of elements not specifically listed in the class definition to be present, but they place an upper limit on the allowable amount.

rutile, alumina, and apatite. In addition, several other categories were defined to accommodate particles not corresponding to any of the above definitions. For example, a "SILICATES" category was defined to include particles with significant silicon content, yet with the balance of the elements in such proportions that the particle would not fit into either the quartz, kaolinite, or illite categories. A "MISCELLANEOUS" category was provided to include those particles whose composition did not allow them to fall into any of the above-mentioned categories. Further descriptions of the instrumental, statistical, classifying, and processing techniques will be given in another publication being readied for submission.

The AIA procedures classified mineral particles into both size and chemical classes. Area-equivalent diameter was used as the size parameter for data presentation. This measurement is the diameter of a circle with the same area as that measured for the mineral particle. Area-equivalent diameter was used instead of a simple length or width measurement because the outline of mineral particles in coal is often complex enough to render such measurements meaningless. The equivalent diameter measurement allows particles to be classified by the area of the particle and yet reported in terms of a linear dimension. Using available literature values for the specific gravity of the individual minerals, the data were then expressed as the weight fraction of the mineral matter within a given mineral/size category. The weight fraction data were then normalized using the mineral matter content to present the mineralogical estimates on a dry coal basis. Such a presentation provides a common base for comparing the coals before and after processing.

The data are presented in Tables 3a and 3b for the raw and clean Illinois coal respectively, and in Tables 4a and 4b for the raw and clean Pittsburgh coal, respectively. Percent removal of the mineral matter calculated for each mineral phase and size category for both coal samples is presented in Tables 3c and 4c.

## RESULTS AND DISCUSSION

Conventional analytical data, obtained by ASTM procedures and presented in Table 1, indicate that the float-sink separation achieved an 84% and a 56% removal of ash from the Illinois and Pittsburgh coals, respectively, with corresponding decreases in total and pyritic sulfur of 50 and 91% and of 43 and 97%, respectively. If expressed as a reduction in the total mineral matter content, based on using a modified Parr formula (4), the corresponding values are 84% for the Illinois coal and 59% for the Pittsburgh coal.

The AIA data for the Illinois coal (Tables 3a-c) and for the Pittsburgh coal (Tables 4a-c) are much more interesting. Results for the raw Illinois coal show that pyrite, quartz, and two clays (kaolinite and illite in approximately equal proportions) make up the bulk (86%) of the mineral matter. The mineral phases are rather uniformly distributed over the entire range of particle size from less than 4  $\mu\text{m}$  to more than 36  $\mu\text{m}$  in diameter.

Table 3a. AIA results for Illinois No. 6 raw coal (200 x 0 mesh), expressed as weight percent of dry coal.

Mineral Phase	Particle Size						Totals
	<4 $\mu\text{m}$	<7 $\mu\text{m}$	<12 $\mu\text{m}$	<21 $\mu\text{m}$	<36 $\mu\text{m}$	>36 $\mu\text{m}$	
Pyrite	0.71	0.77	1.16	1.44	1.62	1.27	6.97
Kaolinite	0.57	0.67	0.65	0.38	0.36	0.20	2.83
Illite	0.32	0.38	0.42	0.42	0.45	0.47	2.47
Quartz	0.84	0.90	1.02	0.71	0.35	0.61	4.43
Siderite	0.10	0.10	0.08	0.14	0.00	0.00	0.42
Calcite	0.04	0.04	0.14	0.11	0.10	0.41	0.83
Silicates	0.23	0.16	0.17	0.10	0.03	0.19	0.89
Minors	0.02	0.03	0.05	0.01	0.04	0.00	0.15
Miscellaneous	0.05	0.06	0.11	0.03	0.07	0.00	0.32
Totals	2.87	3.12	3.79	3.36	3.02	3.15	19.32

Table 3b. AIA results for Illinois No. 6 coal floated at 1.3 specific gravity, as expressed as weight percent of dry coal.

Mineral Phase	Particle Size						Totals
	<4 $\mu\text{m}$	<7 $\mu\text{m}$	<12 $\mu\text{m}$	<21 $\mu\text{m}$	<36 $\mu\text{m}$	>36 $\mu\text{m}$	
Pyrite	0.19	0.28	0.42	0.21	0.07	0.00	1.17
Kaolinite	0.10	0.16	0.09	0.02	0.03	0.00	0.40
Illite	0.12	0.12	0.11	0.07	0.00	0.00	0.42
Quartz	0.16	0.19	0.18	0.09	0.01	0.00	0.63
Siderite	0.02	0.01	0.01	0.02	0.02	0.00	0.08
Calcite	0.00	0.00	0.00	0.00	0.00	0.00	0.01
Silicates	0.08	0.05	0.03	0.02	0.02	0.00	0.20
Minors	0.02	0.02	0.01	0.00	0.00	0.00	0.04
Miscellaneous	0.04	0.03	0.02	0.00	0.00	0.00	0.08
Totals	0.72	0.86	0.87	0.44	0.16	0.00	3.05

Table 3c. Use of AIA results to estimate percent removal of mineral matter from Illinois No. 6 coal. The numbers are based on values in Tables 3a and 3b before they were rounded off.

Mineral Phase	Particle Size						Totals
	<4 $\mu\text{m}$	<7 $\mu\text{m}$	<12 $\mu\text{m}$	<21 $\mu\text{m}$	<36 $\mu\text{m}$	>36 $\mu\text{m}$	
Pyrite	73	64	64	85	95	100	83
Kaolinite	83	77	86	94	91	100	86
Illite	61	69	74	82	100	100	83
Quartz	81	79	83	88	96	100	86
Siderite	81	85	86	89	--	100	80
Calcite	98	93	97	96	100	100	99
Silicates	67	67	81	77	45	100	77
Minors	38	51	75	100	100	100	72
Miscellaneous	30	58	83	100	100	100	75
Total	75	73	77	87	95	100	84

Table 4a. AIA results for Pittsburgh No. 8 raw coal (200 x 0 mesh), expressed as weight percent of dry coal.

Mineral Phase	Particle Size						Totals
	<4 $\mu\text{m}$	<7 $\mu\text{m}$	<12 $\mu\text{m}$	<21 $\mu\text{m}$	<36 $\mu\text{m}$	>36 $\mu\text{m}$	
Pyrite	0.13	0.62	0.72	0.81	1.00	1.19	4.47
Kaolinite	0.08	0.38	0.24	0.12	0.08	0.04	0.94
Illite	0.06	0.26	0.17	0.14	0.16	0.20	0.98
Quartz	0.03	0.13	0.12	0.08	0.01	0.15	0.52
Siderite	0.02	0.10	0.06	0.08	0.14	0.06	0.46
Calcite	0.00	0.02	0.01	0.02	0.00	0.00	0.05
Silicates	0.01	0.02	0.01	0.02	0.07	0.00	0.13
Minors	0.05	0.22	0.12	0.02	0.00	0.00	0.41
Miscellaneous	0.02	0.10	0.05	0.04	0.02	0.09	0.31
Totals	0.40	1.84	1.50	1.33	1.47	1.72	8.26

Table 4b. AIA results for Pittsburgh No. 8 Coal floated at 1.3 specific gravity, as expressed as weight percent of dry coal.

Mineral Phase	Particle Size						Totals
	<4 $\mu\text{m}$	<7 $\mu\text{m}$	<12 $\mu\text{m}$	<21 $\mu\text{m}$	<36 $\mu\text{m}$	>36 $\mu\text{m}$	
Pyrite	0.21	0.33	0.35	0.08	0.00	0.00	0.96
Kaolinite	0.28	0.45	0.21	0.04	0.00	0.00	0.98
Illite	0.24	0.35	0.11	0.01	0.00	0.00	0.70
Quartz	0.07	0.17	0.07	0.02	0.00	0.00	0.34
Siderite	0.00	0.01	0.01	0.03	0.00	0.00	0.05
Calcite	0.00	0.01	0.01	0.01	0.00	0.00	0.02
Silicates	0.02	0.02	0.01	0.00	0.00	0.00	0.05
Minors	0.02	0.03	0.01	0.04	0.00	0.00	0.10
Miscellaneous	0.06	0.08	0.03	0.00	0.00	0.00	0.17
Totals	0.91	1.43	0.80	0.22	0.00	0.00	3.37

Table 4c. Use of AIA results to estimate percent removal of mineral matter from Pittsburgh No. 8 coal. The numbers are based on values in Tables 4a and 4b before they were rounded off.

Mineral Phase	Particle Size						Totals
	<4 $\mu\text{m}$	<7 $\mu\text{m}$	<12 $\mu\text{m}$	<21 $\mu\text{m}$	<36 $\mu\text{m}$	>36 $\mu\text{m}$	
Pyrite	--	46	51	89	100	100	78
Kaolinite	--	--	12	64	100	100	--
Illite	--	--	37	96	100	100	28
Quartz	--	--	39	75	100	100	35
Siderite	76	95	80	63	100	100	88
Calcite	80	62	41	69	100	100	60
Silicates	--	--	32	100	100	100	62
Minors	57	86	89	--	100	100	75
Miscellaneous	--	17	42	100	100	100	45
Total	--	22	46	83	100	100	59

In the supercleaned Illinois coal, the levels of almost all minerals have been reduced by 60-85% somewhat uniformly for most of the particle size ranges up to 21  $\mu\text{m}$ . Above that size, the removal increases sharply, with nearly complete removal for the larger sizes. The 21  $\mu\text{m}$  cut-off seems reasonable for pyrite when the following calculation is considered. A 21  $\mu\text{m}$  pyrite phase (specific gravity of 5.0) embedded in a 200 mesh (i.e. 75  $\mu\text{m}$ ) coal particle (specific gravity of 1.25) accounts for only 2.2% of the total volume. However, because it accounts for 8.2% of the total weight of such a particle, it increases the specific gravity of the total assemblage to over 1.33; it is thus rejected in the float-sink separation. A 12  $\mu\text{m}$  pyrite phase, however, accounts for about 1% of the volume of the 74- $\mu\text{m}$  assemblage, while the specific gravity is 1.29; thus it is carried along with the float fraction. Of course, since the size consist is 200 mesh x 0, there are many small pyrite particles completely liberated or attached to smaller sized coal particles; however, the cut-off at 21  $\mu\text{m}$  is still remarkably sharp. Furthermore, since the specific gravities for the other minerals are lower than for pyrite, similar calculations for the other minerals should yield a higher value for the particle size at the cut-off point.

When the overall removal of 84% of the total mineral matter is broken down by particle size, it shows an expected trend of increasing with increasing particle size. The cleaning effectiveness ranges from about 75% for the smaller particles (4-12  $\mu\text{m}$ ) to 87% at 21  $\mu\text{m}$  and to 100% for particles larger than 36  $\mu\text{m}$ .

The AIA results for the Pittsburgh No. 8 coal (Tables 4a-c) show a distinctly different character of this coal. The mineral matter in the raw sample is more than 50% pyrite. Although the total amount of clay plus quartz is only one-fourth that of the Illinois No. 6 coal, the clay-to-quartz ratio is much higher. In addition, a substantially larger fraction of the mineral content is present in the larger size ranges. This coal exhibits a much sharper cut-off in the effectiveness of cleaning as a function of particle size. The superclean coal fraction contains no particles larger than 21  $\mu\text{m}$  in diameter. The smaller size fractions show only small amounts of mineral matter removed. For the two smallest sizes, even a slight enrichment can be seen for several mineral phases. This apparent enrichment is partially the result of a mathematical anomaly of the normalization process and partially due to the decrease in the total weight by the removal of other mineral matter from the larger fractions. In any case, the absolute mineral content in these very small size fractions is so minute that the allowed analytical errors can account for the discrepancy, and the overall differences become negligible.

Another observation, however, should be scrutinized more closely. In both coals, the content of pyrite as determined by AIA is consistently higher than that calculated from the pyritic sulfur values obtained by ASTM. Since the accuracy (reproducibility between different labs) for ASTM analyses can be within 0.30 or 0.40% (for less than or more than 2.0% pyritic sulfur, respectively)(5), part of the discrepancy could be explained by the possible analytical error. A more plausible interpretation can be given if the ASTM leachings with nitric acid

do not remove all of the pyrite. Significant residues of pyrite have been observed previously in such leached coal samples (6). Another possible explanation can reside in the specific gravity of pyrite used. Although 5.0 is the literature value for mineral-grade pyrite, past work in this laboratory established that coal-derived pyrite can have a specific gravity significantly lower, ranging from about 3.60 for hand-picked samples to 4.25-4.50 for samples which have been extensively cleaned with hot hydrochloric acid (7). Although the absolute values for the pyrite content are divergent, the relative amounts removed (on a wt. % basis) are relatively comparable for both methods of analysis. However, further work is in progress to resolve this issue.

#### CONCLUSIONS

Automated image analysis used in conjunction with scanning electron microscopy and energy-dispersive x-ray analysis has been shown to be an effective tool to characterize in-situ the mineral matter in raw and cleaned coal. Both mineral phase analysis and particle size distribution were obtained for two coals (200 mesh x 0) before and after processing. For Illinois No. 6 coal, which contained mostly pyrite, quartz, kaolinite, and illite rather uniformly distributed among the various particle sizes, the cleaning effectiveness increased gradually with increasing particle size of the mineral phases. The levels of removal ranged from about 75% for the smallest particles to 100% for particles larger than 36  $\mu\text{m}$ . For the Pittsburgh coal, more than half of the mineral matter was pyrite, and the pyrite was relatively coarse. The other minerals were smaller in size. During cleaning of the Pittsburgh coal, most of the large-sized mineral matter was removed, while the finer-sized mineral matter was relatively untouched. The cut-off size was approximately 21  $\mu\text{m}$ .

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#### REFERENCES

1. T. D. Wheelock and R. Markuszewski, "Coal Preparation and Cleaning," Chapter 3 in The Science and Technology of Coal and Coal Utilization, ed. by B. R. Cooper and W. A. Ellingson, Plenum Press, New York, 1984, pp. 47-123.
2. F. E. Huggins, B. A. Kosmack, G. P. Huffman, and R. J. Lee, "Coal Mineralogies by SEM Automated Image Analysis," Scanning Electron Microscopy/1980, Vol. I, pp. 531-540.
3. J. Lebedzik, LeMont Scientific, Inc., State College, PA, personal communication, Fall 1980.
4. P. H. Given and R. F. Yarzab, "Analysis of the Organic Substance of Coals: Problems Posed by the Presence of Mineral Matter," Chapter 20 in Analytical Methods for Coal and Coal Products, Vol. II, ed. by C. Karr, Jr., Academic Press, New York, 1978, pp. 3-41.
5. American Society for Testing and Materials, Annual Book of ASTM Standards, Section 5: Petroleum Products, Lubricants, and Fossil Fuels, Philadelphia, 1983, Method D2492.
6. R. T. Greer, "Organic and Inorganic Sulfur in Coal," Scanning Electron Microscopy/1979, Vol. I, pp. 477-485.
7. K. C. Chuang, M.-C. Chen, R. T. Greer, R. Markuszewski, Y. Sun, and T. D. Wheelock, "Pyrite Desulfurization by Wet Oxidation in Alkaline Solutions," Chem. Eng. Commun. 7, 79-94 (1980).